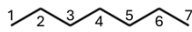
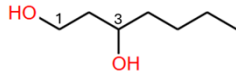
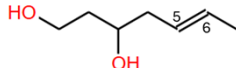
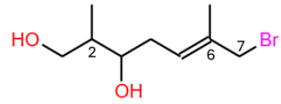
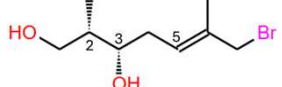


How to Deduce the Structure of an Organic Compound from its Name

(Try the problems at <https://cheminteractive.ie/nomen1.php>; you will get feedback based on the strategy below.)

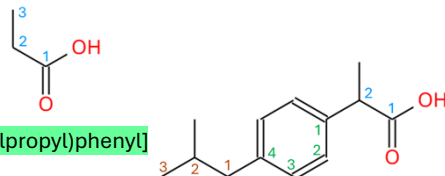
Systematic names contain all the information needed to deduce the corresponding structures, and they are constructed in a consistent way with the constituent parts always in the same order. More than one nomenclature system is in use, and non-systematic names are also common, and this simplified summary only deals with the IUPAC nomenclature system, as exemplified by the name below. The components of IUPAC names are described beneath the example, together with [the strategy for deducing the corresponding structures](#). A basic knowledge of alkane and cycloalkane nomenclature is assumed. Two additional examples follow at the foot of the page, and there are several more on page 2.

(2R,3S,E)-7-bromo-2,6-dimethylhept-5-en-1,3-diol

1. Draw the parent compound.	<p>Most IUPAC names are based on the concept that the structures are modifications of 'parent' compounds, which are almost always unbranched chains or unbranched rings. A name can only contain one parent. The first step in constructing the structure is to draw the parent chain or ring.</p> <p>As well as alkanes, and cycloalkanes, common parents include the rings listed in Tables 1 and 2 on page 2. Some parent compounds include a 'principal functional group' (see below) as well as a chain or ring. This category includes some very common parents, e.g. acetic acid (CH₃CO₂H), see Table 3 for important examples. Some non-systematic parent names are retained for historical reasons, e.g. styrene (C₆H₅CH=CH₂).</p>	<p>Hept means that the parent is heptane. Number the carbons from one (either) end.</p> 	
2. Add the principal functional group(s) to the parent.	<p>If the name ends with anything other than -ane, -ene or -yne, a 'principal functional group' is present. Some common functional groups (FGs) that can be listed as suffixes at the end of compound names are shown in Table 4 on page 2. Note that some FGs are named differently if the parent compound is a ring because they include a carbon that can have only one additional bond, so they must be external to the ring. If the compound is acyclic, those FGs must be at the end of the parent chain. 'Locants' give the position(s) of the FG(s) on the parent, and 'multiplicative prefixes' (di-, tri-, etc.) indicate more than one instance of the principal FG.</p> <p>When present, a principal FG 'controls' the numbering of the parent. For rings, the carbon bearing the FG is numbered 1. For chains that include a principal FG that must be at the end of the chain, the carbon that is included in the FG is numbered 1, e.g. propanal (CH₃CH₂CHO) is numbered from the aldehyde carbon. In these cases, no locant is provided for the principal FG (it is position 1 by definition). The second step is to add the principal FG(s) at the correct position(s).</p>	<p>1,3-diol means that the principal FG is an alcohol, and there are two, at positions 1 and 3.</p> 	
3. Add the unsaturation to the parent.	<p>If the ending of the name of the parent is not -an(e), carbon-carbon double or triple bonds are present, and the next step is to add them to the structure.</p> <p>If the ending is -en(e), a C=C double bond is present, and the locant is the number of the first C of the double bond (e.g. hept-5-en indicates a double bond between carbons 5 and 6 of the parent chain. Likewise, -yn(e) indicates the presence of a C≡C triple bond.</p> <p>'Multiplicative prefixes' indicate that more than one double or triple bond, e.g. deca-1,3,5-triene-9-yne is a 10-carbon chain with double bonds at positions 1, 3 and 5, and a triple bond at position 9.</p>	<p>5-en means that there is one double bond (and no triple) and it is between carbons 5 and 6,</p> 	
4. Add the substituents to the parent and/or the principal FG.	<p>Next, add the <i>substituents</i>, additional groups that are attached to the parent chain or ring. Names of substituents (e.g. 7-bromo) on the parent chain/ring, with <i>locants</i> (e.g. 7-bromo) specifying their locations, are given as prefixes to the parent name. <i>Multiplicative prefixes</i> (e.g. dimethyl) mean that a substituent is present more than once.</p> <p>In addition to alkyl, and cycloalkyl, groups, and the substituents listed in Tables 1 - 4, common substituents include: vinyl (CH₂=CH-), isopropyl ((CH₃)₂CH-), <i>tert</i>-butyl ((CH₃)₃C-), phenyl (C₆H₅-), and benzyl (C₆H₅CH₂-). Alkyloxy groups: methoxy (CH₃O-), ethoxy (CH₃CH₂O-), propoxy (C₃H₇O-), butoxy (C₄H₉O-), pentyloxy (C₅H₁₁O-), etc.</p>	<p>Fluoro (F-), chloro (Cl-), bromo (Br-), iodo (I-), nitro (O₂N-). The functional groups listed in Table 4 on Page 2 can be present as substituents, rather than <i>principal</i> FGs, in which case they may be named quite differently; see the prefixes listed in the Table.</p> <p>'Compound substituted' substituents are ones in which a substituent is itself substituted, e.g. chloromethyl is ClCH₂-. Work from right to left, drawing the substituent, numbering it from the point of attachment to the parent, then doing the same with its substituent, see ibuprofen.</p>	<p>7-bromo-2,6-dimethyl means that there is a bromo (Br) at position 7, and methyl groups (CH₃) at positions 2 and 6.</p> 
5. Add the stereochemistry at chiral centres and/or alkenes.	<p>The final step in drawing the structure is to add the stereochemistry. 'Stereo-descriptors' at the start of the name give the configurations at chirality centres (R or S) and at alkenes (E or Z), according to the Cahn-Ingold-Prelog (CIP) rules (see this excellent resource). <i>Cis</i> and <i>trans</i> are sometimes used to specify the stereochemistry of di-substituted alkenes, and the relative stereochemistry of di-substituted aliphatic rings. Locants (e.g. 2R) specify the positions on the parent chain/ring that the descriptors refer to. Check the <i>locants</i> given in the <i>stereo-descriptors</i> and ensure that the correct configuration is drawn.</p>	<p>(2R,3S,E) means that the chirality centres at positions 2 and 3, and the alkene, have the <i>R</i>, <i>S</i> and <i>E</i> configurations, respectively. The structure is now complete. (2R,3S,E)-7-bromo-2,6-dimethylhept-5-en-1,3-diol is:</p> 	

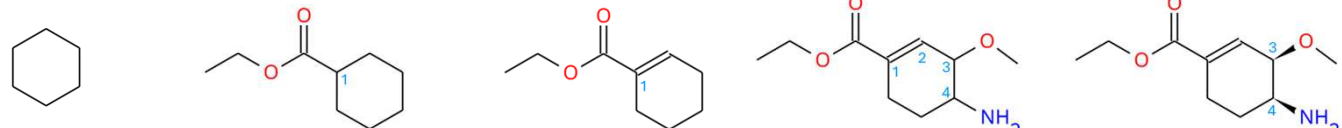
2-[4-(2-Methylpropyl)phenyl]propanoic acid (ibuprofen)

1 Parent **prop**
 2 Principal FG **oic acid**
 3 Unsaturation **an**
 4 Substituent **2-[4-(2-Methylpropyl)phenyl]**



Ethyl (3R,4S)-4-amino-3-methoxycyclohex-1-enecarboxylate

1 Parent **cyclohex** 2 Principal FG **Ethyl -carboxylate** 3 Unsaturation **1-ene** 4 Substituents **4-amino-3-methoxy** 5 Stereo-descriptors **3R,4S**

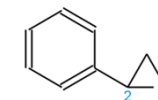
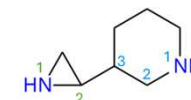
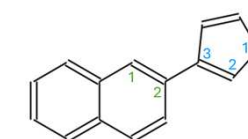


How to Deduce the Structure of an Organic Compound from its Name continued (<https://cheminteractive.ie/nomen1.php>)

These Tables list names of common **parent systems**, and common **principal functional group** suffixes, and also give the prefixes when these parents/functional groups are present as **substituents**.

Table 1: Common Non-Aromatic Heterocyclic Parent Rings

Name	aziridine	1,4-dioxane	morpholine	oxane	oxolane	oxirane	piperidine	piperazine	pyrrolidine
Structure									
Substituent	aziridinyl-	1,4-dioxanyl-	morpholinyl-	oxanyl-	oxolananyl-	oxiranyl-	piperidinyl- (or piperidyl-)	piperazinyl-	pyrrolidinyl-

2-Phenylloxirane

3-(Aziridin-2-yl)piperidine

3-(Naphthalen-2-yl)furan

Table 2: Common Aromatic Parent Rings

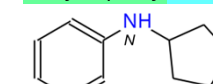
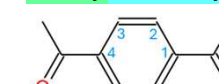
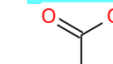
Name	benzene	furan	imidazole	indole	isoquinoline	naphthalene	pyridine	pyrrole	quinoline	thiophene
Structure										
FG prefix	phenyl-	furanyl- (or furyl-)	imidazolyl-	indolyl-	isoquinolinyl- (or isoquinoyl-)	naphthalenyl- (or naphthyl-)	pyridinyl- (or pyridyl-)	pyrrolyl-	quinolinyl- (or quinoyl-)	thiophenyl- (or thienyl-)

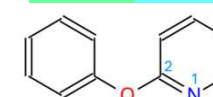
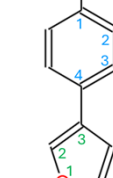
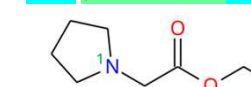
Table 3: Common Functional Parents

Name	formic acid*	acetic acid*	benzoic acid*	phenol	aniline	urea
Structure						
FG prefix	formyl-	acetyl-	benzoyl-	phenoxy-	anilino-	

* Also, the esters (RC(=O)OR'), acid halides (RC(=O)X), amide (RC(=O)NH₂), nitrile (RC≡N), and aldehyde (RC(=O)H)

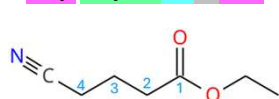
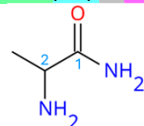
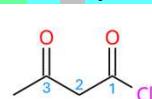
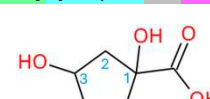
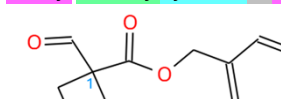
N-Methylformamide

N-Cyclopentylaniline

4-Acetylbenzaldehyde

4-(Furan-3-yl)-benzoic acid

3-Bromo-4-nitrophenol

2-Phenoxy pyridine

Ethyl (pyrrolidin-1-yl)acetate

Table 4: Common Functional Groups

Class	Carboxylic acids	(Carboxylic) Esters	Acid halides	Amides	Nitriles	Aldehydes	Ketones	Alcohols	Amines
FG suffix	-oic acid -carboxylic acid	R -oate R -carboxylate	-oyl halide -carbonyl halide	-amide -carboxamide	-nitrile -carbonitrile	-al -carbaldehyde	-one	-ol	-amine
Formula	-C*(=O)OH	-C*(=O)OR	-C*(=O)X	-C*(=O)NH ₂	-C≡N	-C*(=O)H	=O	-OH	-NH ₂
Examples	CH ₃ CO ₂ H Acetic acid C ₃ H ₅ CO ₂ H Cyclopropane-carboxylic acid	CH ₃ CH ₂ CO ₂ CH ₃ Methyl propanoate C ₆ H ₅ CO ₂ CH ₂ CH ₃ Ethyl cyclopentanoate	CH ₃ CH ₂ CH ₂ COCl Butanoyl chloride C ₄ H ₇ COCl Cyclobutane-carbonyl chloride	CH ₃ CH ₂ CONH ₂ Propanamide C ₆ H ₁₁ CONH ₂ Cyclohexane-carboxamide	CH ₃ CH ₂ CH ₂ C≡N Butanenitrile C ₄ H ₇ C≡N Cyclobutane-carbonitrile	CH ₃ CHO Acetaldehyde (ethanal) C ₃ H ₅ C(=O)H Cyclopropane-carbaldehyde	CH ₃ C(=O)CH ₂ CH ₂ CH ₃ Pentan-2-one	CH ₃ CH(OH)CH ₃ Propan-2-ol	C ₄ H ₇ -NH ₂ Cyclobutanamine
FG prefix	carboxy-	alkoxycarbonyl-	halocarbonyl-	carbamoyl-	cyano-	oxo- (chain), formyl- (ring)	oxo-	hydroxy-	amino-

* If the parent is a chain, this C is included in the name, e.g. CH₃CH₂CO₂H is propanoic acid. If the parent is a ring, this C is external to the ring and is not included in the parent name, e.g. C₆H₅CO₂H is cyclohexanecarboxylic acid.

Ethyl 4-cyanobutanoate

2-Aminopropanamide

3-Oxobutanoyl chloride

1,3-Dihydroxycyclopentanecarboxylic acid

Benzyl 1-formylcyclobutanoate

5-Hydroxyhexan-2,4-dione
